

Density of Seawater

Equation of State: $\rho = \rho(T, S, p)$

T = Temperature

units: °C

ocean range: -2°C to 30°C

Potential Temperature

if we raise water without changing heat content

==> pressure decreases

==> temperature decreases

convenient to use *potential temperature* θ

$\theta = T$ water would have at surface

$\theta < T$

difference pretty small....

$\theta \approx T - .5^\circ\text{C}$ for 5 km

BUT important effect for deep water

Accurate formula for θ can be found in Gill

My approx formula: $\theta = T - Ap - Bp^2$

$A = .04(1 + .185T + .035[S - 35])$

$B = .0075(1 - T/30)$

T in deg C, p in “dynamic km”

For $30 \leq S \leq 40$, $-2 \leq T \leq 30$, $p \leq 6$ km,

$\theta - T$ good to about 6%

(except for some shallow values w/ tiny $\theta - T$)

S = Salinity = mass of salt (gm) dissolved in 1 kg seawater
units: ppt or psu

kind of salt the same everywhere in sea

(ocean mixes up in time it takes to dissolve)

55% chlorine

31% sodium

8% sulphate

4% magnesium

1% potassium

S can go from 0 (coast) to about 40psu (Red Sea)

BUT 90% is between 34 and 35 psu

P = pressure = force/area

for ocean, use *hydrostatic* approximation:

pressure = weight of fluid above

MKS units: Newton/m² = Pascal

atmospheric weight $\approx 10^5$ N/m² \equiv 1 bar

ocean weight \approx 1 bar every 10 m depth of water

\implies ocean depth in dbar \approx depth in meters

ocean pressure measured relative to surface (ignore atm)

Rules About Density

- 1) Density hardly changes

$$\rho \approx 1000 \text{kg/m}^3 = 1 \text{ gm/cm}^3$$

- 2) Density increases when pressure increases

z	ρ	$(S = 35 \text{psu}, T = 0^\circ\text{C})$
0	1028.1	
1000	1032.8	
4000	1046.4	

- 3) Density increases when salinity increases

S	ρ	$(T = 0, z = 0)$
0	999.8	
35	1028.1	

- 4) Density decreases as T increases

- 5) BUT the *amount* it changes depends on T, p

colder water: *less* sensitive to T

deeper water: *more* sensitive to T

- 6) Variations due to T and S are small but **very important**

Accurate formula for $\rho(T, S, p)$ can be found in Gill appendix

My Approximate Density Formula:

$$\rho = C(p) + \beta(p)S - \alpha(T, p)T - \gamma(T, p)(35 - S)T$$

units: p in “km”, S in psu, T in $^\circ\text{C}$

$$C = 999.83 + 5.053p - .048p^2$$

$$\beta = .808 - .0085p$$

$$\alpha = .0708(1 + .351p + .068(1 - .0683p)T)$$

$$\gamma = .003(1 - .059p - .012(1 - .064p)T)$$

For $30 \leq S \leq 40$, $-2 \leq T \leq 30$, $p \leq 6$ km:

good to $.16 \text{ kg/m}^3$

For $0 \leq S \leq 40$, good to $.3 \text{ kg/m}^3$

cabelling

nonlinear T dependence, so mixing changes density

thermobaric effects

α depends on p , so

density ordering depends on depth

Potential Density

sigma-t: $\sigma_t = \rho(T, S, 0) - 1000$

potential density: $\rho(\theta, S, 0)$

$\sigma_\theta = \rho(\theta, S, 0) - 1000$

$\sigma_1 = \rho(\theta, S, 1\text{km}) - 1000$